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Write net ionic equations for the following acid base reactions. H_2SO_4 and $\text{Ba}(\text{OH})_2$. HCl and Na_2CO_3 (Carbonic acid formula Study Guide:Acids and Bases

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ACIDS AND BASES STUDY GUIDE NAME _____#_____ 1. Label the acid, label the base, and label the conjugate acid-base pair. $\text{NH}_3 + \text{H}_2\text{O} \rightleftharpoons \text{NH}_4^+$

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(assuming that there is a reaction)? Acids and bases break down into charge ions called _____ that can conduct Chapter 18 and 19 Study Guide

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